

## Quiz & Worksheet - Predicting the Entropy of Physical and Chemical Changes

1. Which of the following substances has the highest entropy at 25°C?

- ☐  $\text{CH}_3\text{OH}(l)$
- ☐  $\text{CO}(g)$
- ☐  $\text{MgO}(s)$
- ☐  $\text{CaCO}_3(s)$

2. How is the net change in entropy calculated?

- ☐  $\Delta S = S_{\text{final}} - S_{\text{initial}}$
- ☐  $\Delta S = S_{\text{initial}} - S_{\text{final}}$
- ☐  $\Delta S = S_{\text{final}} + S_{\text{initial}}$
- ☐  $\Delta S = S_{\text{final}} / S_{\text{initial}}$

3. The order of entropy levels in different phases from lowest to highest is:

- ☐ Solid, gas, liquid
- ☐ Liquid, gas, solid
- ☐ Solid, liquid, gas
- ☐ Gas, liquid, solid

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**Bernd Finkbeiner, Leonardo Mariani**



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supporting the Kremlin's ideology Ensuring cyber safety from ransomware attacks has become a national security priority for many nations across the world The evolving variants of ransomware attacks present a wider and more challenging threat landscape highlighting the need for collaborative work throughout the entire cyber ecosystem value chain In response to this evolving threat a book addressing the challenges associated with ransomware is very timely This book aims to provide a comprehensive overview of the evolution trends techniques impact on critical infrastructures and national security countermeasures and open research directions in this area It will serve as a valuable source of knowledge on the topic

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