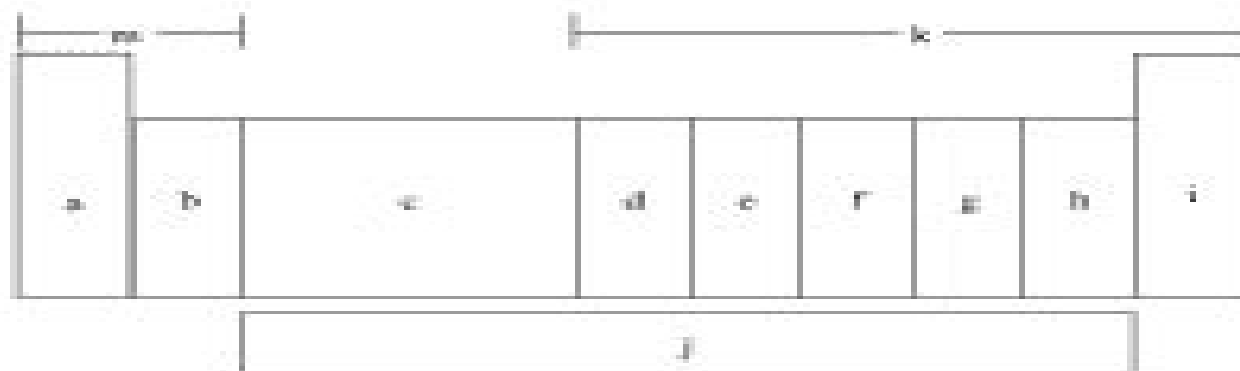


5-2 Review and Reinforcement

Reading the Periodic Table

On the line at the left, write the letter of the approximate location of each group of elements on the periodic table below. Some letters will be used more than once.

- | | |
|----------------------------------|---|
| _____ 1. carbon family | _____ 8. f-block elements |
| _____ 2. alkaline earth metals | _____ 9. noble gases |
| _____ 3. inner transition metals | _____ 10. p-block elements |
| _____ 4. halogens | _____ 11. nitrogen family |
| _____ 5. d-block elements | _____ 12. s-block elements |
| _____ 6. oxygen group | _____ 13. transition metals |
| _____ 7. alkali metals | _____ 14. group of one nonmetal and four metals |



Use the skills you developed in Section 5-2 to answer each of the following questions. Below is the abbreviated electron configuration for sodium. Explain each part of this configuration in the space provided.

- $[Ne] 3s^1$
- | |
|-----------|
| 15. _____ |
| 16. _____ |
| 17. _____ |
| 18. _____ |

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Prentice-Hall Staff



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1. Understanding the eBook Falling Fast Reinforcement Answer Guide
 - The Rise of Digital Reading Falling Fast Reinforcement Answer Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Falling Fast Reinforcement Answer Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Falling Fast Reinforcement Answer Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Falling Fast Reinforcement Answer Guide

- Personalized Recommendations
- Falling Fast Reinforcement Answer Guide User Reviews and Ratings
- Falling Fast Reinforcement Answer Guide and Bestseller Lists
- 5. Accessing Falling Fast Reinforcement Answer Guide Free and Paid eBooks
 - Falling Fast Reinforcement Answer Guide Public Domain eBooks
 - Falling Fast Reinforcement Answer Guide eBook Subscription Services
 - Falling Fast Reinforcement Answer Guide Budget-Friendly Options
- 6. Navigating Falling Fast Reinforcement Answer Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Falling Fast Reinforcement Answer Guide Compatibility with Devices
 - Falling Fast Reinforcement Answer Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Falling Fast Reinforcement Answer Guide
 - Highlighting and Note-Taking Falling Fast Reinforcement Answer Guide
 - Interactive Elements Falling Fast Reinforcement Answer Guide
- 8. Staying Engaged with Falling Fast Reinforcement Answer Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Falling Fast Reinforcement Answer Guide
- 9. Balancing eBooks and Physical Books Falling Fast Reinforcement Answer Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Falling Fast Reinforcement Answer Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Falling Fast Reinforcement Answer Guide
 - Setting Reading Goals Falling Fast Reinforcement Answer Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Falling Fast Reinforcement Answer Guide

- Fact-Checking eBook Content of Falling Fast Reinforcement Answer Guide
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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