

## Pre-Lab Information

<b>Purpose</b>	Explore the relationship between temperature and solubility using a laboratory procedure.
<b>Time</b>	Approximately 45 minutes
<b>Question</b>	What is the effect of temperature on the solubility of a solid in a liquid?
<b>Hypothesis</b>	If the temperature of the liquid is increased, then more sugar will dissolve because warm solutions hold more solute than cold solutions.
<b>Variables</b>	<i>Independent Variable:</i> temperature of water <i>Dependent Variable:</i> amount of sugar that can be dissolved
<b>Summary</b>	You will measure the amount of sugar you can dissolve at four (or more) different temperatures of water and analyze the resulting data.

## Safety

- Always wear a lab gown and safety goggles while performing an experiment.
- Behavior in the lab needs to be purposeful. Use caution while working around the hot plate.
- Check glassware, such as your beaker, for cracks and chips prior to use.
- Remember to always use the tongs while handling the hot beaker.
- Report all accidents—no matter how big or small—to your teacher.

## Procedure

### Step 1: Gather Materials

- 100 mL beaker
- beaker tongs
- spoon
- stirring rod
- thermometer
- hot plate
- hot pad
- balance
- 300 g sugar
- ice
- water

### Step 2: Measure the Mass of One Spoonful of Sugar

Use a weighing paper to measure the number of grams of sugar that are in one spoonful of sugar. Be sure to use a level spoonful, not a heaping spoonful, to help ensure that each time you add sugar, it is approximately the same amount. Record that value above the data table.

### Step 3: Measure Solubility in Cold Water

- Place ice in a 100 mL beaker. Fill the beaker with water until the ice and water total 50 mL. Stir until just a small amount of ice remains. Let the temperature of the chilled water stabilize to obtain a temperature  $\approx 5^{\circ}\text{C}$  (between  $0^{\circ}\text{C}$  and  $10^{\circ}\text{C}$ ).
- Record the temperature in  $^{\circ}\text{C}$ .
- Add a spoonful of sugar (again, a level spoonful, not a heaping spoonful). Stir using the stirring rod. Continue adding spoonfuls of sugar until the solution becomes saturated, stirring after each spoonful is added.
- After the solution reaches saturation, record the amount of sugar (in spoonfuls) dissolved before the saturation point. **Do not discard the solution; continue to use it throughout the rest of the procedure.**

# Fcps Student Lab Guide

**Paula Olszewski-Kubillus, Rena F.  
Subotnik, Frank C. Worrell**



## **Fcps Student Lab Guide:**

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**managerial finance 13th edition by lawrence gitman** - Apr 16 2022

web nov 25 2022 book name principles of managerial finance authors lawrence j gitman chad j zutter edition 13th edition

language english file size 14 10 mb pages 947 media source google drive

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