Name \_\_\_\_\_

#### Elements, Compounds & Mixtures Worksheet

Part 1: Read the following information on elements, compounds and mixtures. Fill in the blanks where necessary.

#### Elements:

- A pure substance containing only one kind of <u>atom</u>
- An element is always uniform all the way through (homogeneous).
- An element <u>cannot</u> be separated into simpler materials (except during nuclear reactions).
- Over 100 existing elements are listed and classified on the <u>Periodic Table</u>.

#### Compounds:

- A pure substance containing two or more kinds of <u>atoms</u>.
- The atoms are <u>chemically</u> combined in some way. Often times (but not always) they come together to form groups of atoms called molecules.
- · A compound is always homogeneous (uniform).
- Compounds <u>cannot</u> be separated by physical means. Separating a compound requires a chemical reaction.
- The properties of a compound are usually different than the properties of the elements it contains.

#### Mixtures:

- Two or more <u>elements</u> or <u>compounds</u> NOT chemically combined.
- No reaction between substances.
- Mixtures can be uniform (called <u>homogeneous</u>) and are known as solutions.
- Mixtures can also be non-uniform (called \_\_\_heterogeneous\_\_\_).
- Mixtures can be separated into their components by chemical or physical means.
- · The properties of a mixture are similar to the properties of its components.

Part 2: Classify each of the following as elements (E), compounds (C) or Mixtures (M). Write the letter X if it is none of these.

_E_Diamond (C)	_C_Sugar (C6H12O6)	_M_Milk	_E_Iron (Fe)
_M_Air	C_Sulfuric Acid (H <sub>2</sub> SO <sub>4</sub> )	_M_Gasoline	_X_Electricity
_E_Krypton (K)	_E_Bismuth (Bi)	_E_Uranium (U)	_M_Popcorn
_C_Water (H <sub>2</sub> O)	C_Alcohol (CH <sub>2</sub> OH)	_M_Pail of Garbage	M_A dog
C_Ammonia (NH	b) C Salt (NaCl)	X_Energy	E Gold (Au)
_M_Wood	_M_Bronze	_M_lnk	_M_Pizza
C Dry Ice (CO2)	C Baking Soda (NaHCO)	E Titanium (Ti)	M Concrete

# **Element Soup Worksheet Answers**

R Bogdan

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