

ENERGY CHANGES AND RATES OF REACTIONS

THERMOCHEMISTRY

Thermochemistry is the study of the quantity of heat energy released or absorbed in a chemical reaction.

Energy can be defined as the ability to do work.

Work is the energy transferred to an object by a force that causes the object to move.

Potential energy – is the energy a body has due to its position or its composition. It is a stored energy.

Kinetic energy – is the energy a body has due to its motion.



Internal energy – is the sum of all potential and kinetic energy.

All forms of energy can be classified as either potential or kinetic energy.

The energy associated with chemical bonds is potential energy.

Total energy is equal to internal energy.

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Jules Michael Kline



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