

## Explaining Substitution

Why?

**Answer:** The groups off that side of the stereocenter that don't react with the nucleophile aren't involved.

**Example:** 2-bromo-2-methylbutane is reacted with an alkoxide nucleophile. The nucleophile, not protons H<sup>+</sup> and OH<sup>-</sup>, determines stereochemistry.

### Observations

- Secondary alkyl halides: SN2 50%, SN1 50%
- Tertiary alkyl halides: SN1 100%, SN2 0%
- Methyl halides: SN2 100%, SN1 0%
- Primary alkyl halides: SN2 100%, SN1 0%
- Tertiary alkyl halides: SN1 100%, SN2 0%
- Tertiary, quaternary ammonium salts: SN2 0%, SN1 100%
- Solvent
- Solvent polarity
- Solvent PDI value
- Solvent PDI: SN2 vs SN1 value
- Solvent hydrogen bonding
- Solvent proticity
- Size

# Exploring Equilibrium Minilab

**S Marginson**



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